

Coenzyme B Induced Coordination of Coenzyme M via Its Thiol Group to Ni(I) of F₄₃₀ in Active Methyl-Coenzyme M Reductase

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Methane is formed in methanogenic archaea by the reduction of methyl-coenzyme M (CH₃-S-CoM) with coenzyme B (HS-CoB) to CH₄ and the heterodisulfide CoM-S-S-CoB.¹ This reaction is catalyzed by methyl-coenzyme M reductase (MCR), which is composed of three different subunits in an α₂β₂γ₂ arrangement and which contains tightly bound 2 mol of the nickel porphyrinoid F₄₃₀. Crystal structures of the 300 kDa enzyme with and without coenzymes or product bound have been resolved to 1.16 Å.² They were, however, only obtained for the enzyme in the inactive Ni(II) state. For the enzyme to be active, the prosthetic group has to be in the Ni(I) oxidation state,^{3–5} which is rapidly lost by autoxidation of the Ni(I).⁶ Therefore, it has not been known until now how within the active enzyme the active site Ni(I) interacts with the substrates.

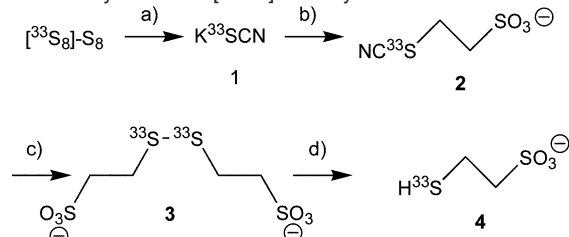
Active MCR exhibits an axial EPR signal MCR_{red1} derived from Ni(I) which does not change significantly when CH₃-S-CoM alone or together with HS-CoB are added to the active enzyme, showing that although an enzyme-substrate-complex is formed, there is no detectable direct interaction of the substrates with the Ni(I) of the prosthetic group.⁷ The XAS data of the active enzyme are also the same in the absence and presence of the substrates.⁸

The reduction of CH₃-S-CoM catalyzed by MCR is inhibited by coenzyme M (HS-CoM), inhibition being reversible and competitive to CH₃-S-CoM. In the presence of only HS-CoM, the enzyme shows the axial MCR_{red1} EPR signal. In the presence of both HS-CoM and HS-CoB, however, the axial signal is partially converted into the highly rhombic EPR signal MCR_{red2}.⁷ On the basis of ¹H and ¹⁴N data obtained from electron nuclear double resonance (ENDOR) and hyperfine sublevel correlation spectroscopy (HYSCORE) measurements, it was proposed that in the MCR_{red2} state HS-CoM is axially coordinated to Ni(I).⁹ Here we report experiments with ³³S-labeled HS-CoM, proving that the thiol group of HS-CoM coordinates to the Ni(I) ion of F₄₃₀.

[2-³³S]-coenzyme M (**4**) (Scheme 1) was synthesized in a one-pot procedure starting from elemental sulfur ([³³S₈]-S₈)¹⁰ and potassium cyanide¹¹ according to Scheme 1.¹²

Figure 1 shows the X-band EPR spectra of MCR_{red2}-HS-CoM (³²S (99.25%) with nuclear spin *I* = 0, ³³S (0.75%) with *I* = 3/2) and MCR_{red2}-H³³S-CoM. For a better comparison, the signals of MCR_{red1} were subtracted from the red1/red2 mixture normally shown by these preparations. The EPR spectrum of ³³S-labeled MCR_{red2} shows a pronounced line broadening at the high-field feature corresponding to the *g*₃ principal value. This is a strong indication for the presence of a large ³³S hyperfine coupling along this principal axis direction. No significant broadenings are observed

Scheme 1. Synthesis of [2-³³S]-Coenzyme M^a



^a Conditions: (a) N₂, KCN, EtOH, 4 h, reflux. (b) N₂, BrCH₂CH₂SO₃Na, DMF, 4 h, 120 °C. (c) (1) N₂, K₂CO₃, H₂O, 1 h, 60 °C; (2) H⁺, NH₃. (d) N₂, DTT, H₂O, 30 min. Overall yield from [³³S₈]-S₈, 87% (NMR); after final purification for MCR-assay, 44%.

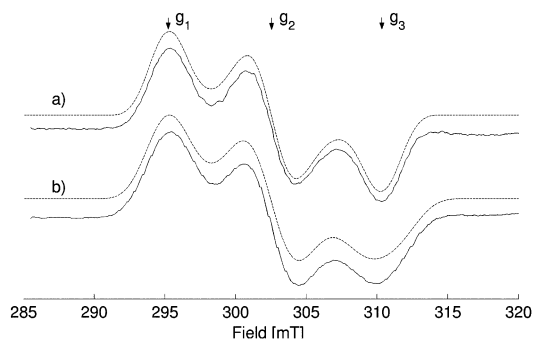


Figure 1. X-band EPR spectra of MCR in the MCR_{red2} state (coenzyme M inhibited enzyme in the presence of coenzyme B). (a) With HS-CoM. (b) With H³³S-CoM. Solid lines: experimental spectra after subtraction of MCR_{red1} signals. Dashed lines: simulations, spectrum b used the same spin Hamiltonian parameters as in (a) but with the addition of a ³³S hyperfine interaction, |A_{1,2}| = 20 MHz, |A₃| = 35 MHz. Experimental conditions: 77 K, modulation amplitude 0.6 mT, microwave frequency 9.45 GHz (Supporting Information contains further details).

at *g*₁ and *g*₂. From spectral simulations, the ³³S hyperfine coupling along *g*₃ is estimated to be roughly |A₃| = 35 MHz, with upper limits along *g*₁ and *g*₂ of |A_{1,2}| = 25 MHz.

In contrast to MCR_{red2}, when either the ox1 or the red1 form of MCR is incubated in the presence of H³³S-CoM, no significant line broadenings are observed in the EPR spectra. This shows that any interaction with ³³S in these two forms is small (as compared to the EPR spectral resolution).

A clear-cut proof of the coordination of HS-CoM to Ni(I) is obtained from HYSCORE spectra measured at Q-band¹³ at the low-field end (*g*₁ value) of the EPR spectrum. Figure 2a and 2b shows the single-crystal-like HYSCORE spectra of MCR_{red2}-HS-CoM and MCR_{red2}-H³³S-CoM at *g*₁ which are free of contributions from other paramagnetic species of MCR. The additional peaks observed in Figure 2b originate from ³³S interactions (labeled in Figure 2b). The two cross-peaks in the (– +)-quadrant at (–10.8, 31.8) MHz and (–31.8, 10.8) MHz are assigned to triple-quantum transitions

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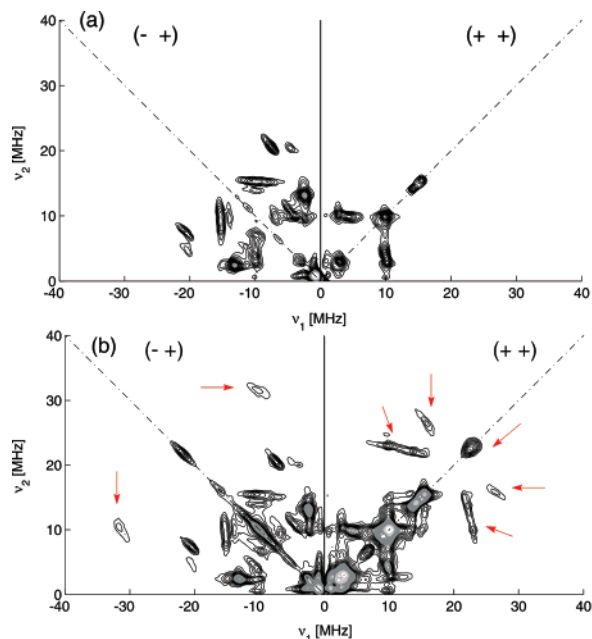


Figure 2. Q-band HYSCORE spectra recorded at 25 K of MCR in the MCR_{red2} state; observer position at g_1 . (a) With HS-CoM. (b) With $H^{33}S$ -CoM. The arrows identify peaks that originate from ^{33}S interactions. The Supporting Information contains further details.

with $\Delta m_1 = 3$. For a simplified system with an isotropic g tensor and an axial hyperfine tensor, the two triple-quantum frequencies can be written to first order as

$$\nu_{TQ}^{(+/-)} = 3 \left[\left(\pm \frac{A_{\perp}}{2} + \nu_1 \right)^2 \sin^2 \beta + \left(\pm \frac{A_{\parallel}}{2} + \nu_1 \right)^2 \cos^2 \beta \right]^{1/2}$$

with the hyperfine principal values A_{\perp} and A_{\parallel} , the nuclear Zeeman frequency ν_1 , and the angle β between the A_{\parallel} principal axis and the static magnetic field vector \mathbf{B}_0 . For a nuclear quadrupole interaction that is small as compared to the hyperfine interaction, these frequencies are to first-order independent of the nuclear quadrupole interaction and differ by $6\nu_1$ for $\beta = 0, 90^\circ$. In Figure 2b, the observed splitting of 20.9 MHz is slightly smaller than $6\nu_1 = 21.7$ MHz, indicating that the orientations selected in this experiment are close to the principal axis of A_{\perp} . The ^{33}S hyperfine coupling A along g_1 can easily be estimated from the equation

$$\nu_{TQ}^{(+)^2} - \nu_{TQ}^{(-)^2} = 18\nu_1(a_{iso} + T(3 \cos^2 \beta - 1)) = 18\nu_1 A$$

where a_{iso} is the isotropic hyperfine coupling, and T is the dipolar coupling.¹⁴ Along g_1 , we then find $|A| = 13.8$ MHz, and for the principal value $|A_{\perp}|$ we estimate a coupling of about 15 MHz.

Several additional peaks are observed in the (+ +)-quadrant of Figure 2b. The strong diagonal peak at 23 MHz is most probably a sulfur double-quantum transition ($\Delta m_1 = 2$), and the cross-peaks represent correlations between sulfur transitions and/or nitrogen-sulfur combination transitions. An unequivocal assignment of all of the new peaks observed in the ^{33}S sample is difficult because the HYSCORE spectrum could only be observed along g_1 . This is because the large anisotropy of the ^{33}S hyperfine coupling broadens the peaks beyond detection as soon as the B_0 observer field in the HYSCORE experiments is shifted to higher values.

The combination of EPR and HYSORE data proves that, in the MCR_{red2} state, HS-CoM is directly coordinated to the Ni(I) ion. An estimate of the spin density on ^{33}S can be obtained from the hyperfine tensor and by considering the relative signs of the principal values. Assuming three positive principal values $[A_1, A_2, A_3] = (15, 15, 35)$ MHz yields an isotropic part of $a_{iso} = 21.7$ MHz and a dipole part of $(-6.7, -6.7, 13.4)$ MHz. For a hyperfine tensor $[A_1, A_2, A_3] = (-15, -15, 35)$ MHz, the isotropic part is $a_{iso} = 1.7$ MHz, and the dipole part is $(-16.7, -16.7, 33.4)$ MHz. In the first case, a spin density of 0.6% in the s-orbital is estimated from the isotropic part, and a spin density of 7% in a 3p-orbital is estimated from the dipolar part.¹⁵ For the second case, the corresponding values are 0.05% (s-orbital) and 17% (3p-orbital). In either case, the large spin density on the sulfur ligand is further proof that the ground state of MCR_{red2} has a relatively high percentage of d_{z^2} character, and the large hyperfine coupling A_3 corroborates the proposal that the g_3 principal axis is perpendicular to the macrocycle.⁹

On the basis of the finding that the HS-CoM interaction is dependent on the presence of HS-CoB in MCR_{red2} , we propose that HS-CoB is not only required as a second substrate,¹⁶ but also to induce a change forcing the real substrate, $CH_3-S-CoM$, and Ni(I) of the prosthetic group to interact in the active enzyme MCR_{red1} .

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Supporting Information Available: Analytical data of $[2-^{33}S]$ -coenzyme M (4, NH_4^+ -form) and experimental procedures (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- Thauer, R. K. *Microbiology* **1998**, *144*, 2377–2406.
- Grabarse, W.; Mahler, F.; Duin, E. C.; Goubeaud, M.; Shima, S.; Thauer, R. K.; Lamzin, V.; Ermler, U. *J. Mol. Biol.* **2001**, *309*, 315–330.
- Goubeaud, M.; Schreiner, G.; Thauer, R. K. *Eur. J. Biochem.* **1997**, *243*, 110–114.
- Telsler, J.; Davydov, R.; Horng, Y. C.; Ragsdale, S. W.; Hoffman, B. M. *J. Am. Chem. Soc.* **2001**, *123*, 5853–5860.
- Tang, Q.; Carrington, P. E.; Horng, Y. C.; Maroney, M. J.; Ragsdale, S. W.; Bocian, D. F. *J. Am. Chem. Soc.* **2002**, *124*, 13242–13256.
- Mahler, F.; Bauer, C.; Jaun, B.; Thauer, R. K.; Duin, E. C. *J. Biol. Inorg. Chem.* **2002**, *7*, 500–513.
- Mahler, F.; Grabarse, W.; Kahnt, J.; Thauer, R. K.; Duin, E. C. *J. Biol. Inorg. Chem.* **2002**, *7*, 101–112. Erratum: *J. Biol. Inorg. Chem.* **2002**, *7*, 351.
- Duin, E. C.; Cosper, N. J.; Mahler, F.; Thauer, R. K.; Scott, R. A. *J. Biol. Inorg. Chem.* **2002**, *8*, 141–148.
- Finazzo, C.; Harmer, J.; Jaun, B.; Duin, E. C.; Mahler, F.; Thauer, R. K.; Van Doorslaer, S.; Schweiger, A. *J. Biol. Inorg. Chem.*, in press.
- $[^{33}S_8]$ -S₈ was purchased from Campro Scientific, Berlin, Germany. The analysis of (isotopic) purity of $[^{33}S_8]$ -S₈ was established by the Kurchatov Institute, Moscow, Russia (isotopic purity: 99.79%; purity: > 99.95%).
- Castiglioni, A. *Gazz. Chim. Ital.* **1933**, *63*, 171.
- During development of the procedure with natural abundance sulfur, each intermediate was isolated and characterized. The product was converted to the ammonium salt by ion exchange with Amberlite, and inorganic salts (NH_4Br) were removed by precipitation of **4** with diethyl ether from a concentrated solution in methanol. The sample used for the experiments with the enzyme was recrystallized from methanol/2-propanol to remove a small inorganic impurity that quenched the MCR_{red2} signal. The details of the development of the synthesis and the analytical data for all intermediates will be published elsewhere.
- Gromov, I.; Shane, J.; Forrer, J.; Rakhmatoullin, R.; Rozentzwaig, Yu.; Schweiger, A. *J. Magn. Reson.* **2001**, *149*, 196–203.
- Schweiger, A.; Jeschke, G. *Principles of Pulse Electron Paramagnetic Resonance*; Oxford University Press: New York, 2001.
- Morton, J. R.; Preston, K. F. *J. Magn. Reson.* **1978**, *30*, 577–582.
- Horng, Y. C.; Becker, D. F.; Ragsdale, S. W. *Biochemistry* **2001**, *40*, 12875–12885.

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